

Ring Enlargement Reaction of 3-Hydroxy-3-propargylisindolin-1-ones : A New Synthetic Method for the 2-Benzazepine-1,5-diones

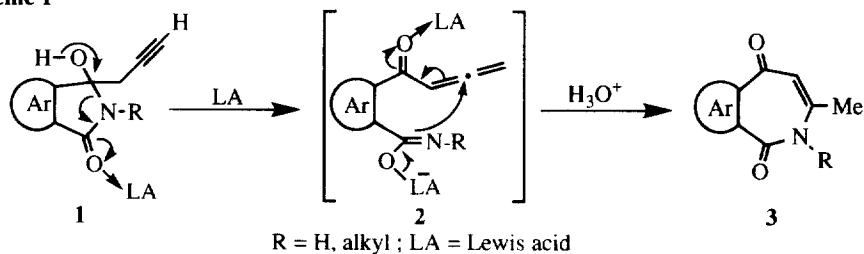
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Abstract : Several 2-benzazepine-1,5-diones were efficiently obtained by treatment of the corresponding *N*-alkyl-3-hydroxy-3-propargylisindolin-1-ones with some tentative Lewis acids *via* intramolecular endo-mode cyclization in the allenyl ketone intermediates generated *in situ*.

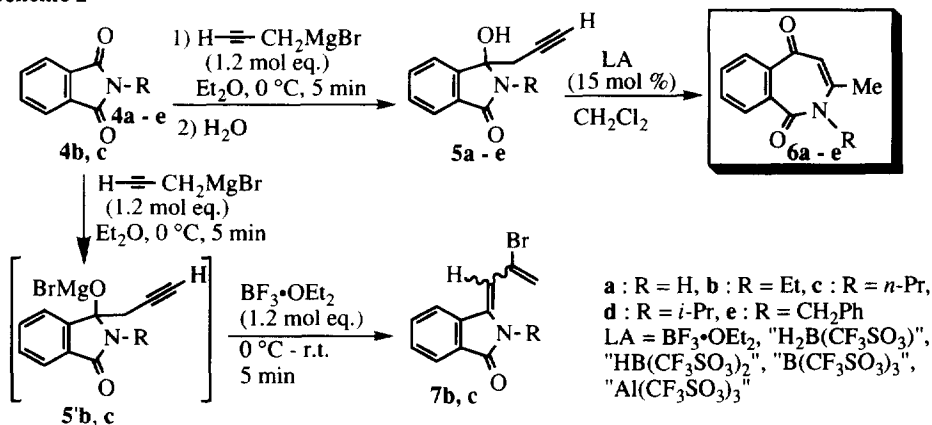
Recently, considerable interest has been focused on the development of new construction methods¹ for the benzazepine skeletons which are common structural moieties of pharmacologically active alkaloids.^{1d,g} We have recently disclosed various carbocyclic endo-mode cyclization reactions by utilizing the intramolecular Michael type addition of the aromatic ring to the conjugated allenyl ketone moiety in the presence of some Lewis acids.² On the basis of our earlier studies,² we designed an expeditious synthetic method for the 3,4-olefinic 2-benzazepine-1,5-diones by utilizing the Lewis acid-promoted ring enlargement³ of hydroxy propargyl γ -lactams **1** toward 7-membered lactams **3** *via* intramolecular endo-mode ring closure in the allenyl ketones **2** generated *in situ* as shown in Scheme 1. Thus, treatment of commercially available phthalimide **4a** and the Gabriel reaction⁴

Scheme 1



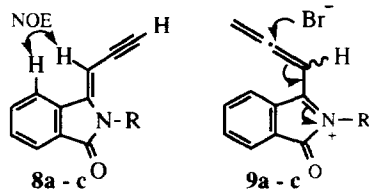
products **4b - e** with 1.2 mol eq. of propargylmagnesium bromide^{2a} in Et₂O at 0 °C for 5 min readily afforded the corresponding *N*-alkyl-3-hydroxy-3-propargylisindolin-1-ones **5a - e** as a colorless powder or needles in 69 - 85% yields (Scheme 2 and Table 1). The structures of **5a - e** were confirmed by their characteristic spectroscopic data [IR (CHCl₃) ν 3295–3272 (–C≡CH), 3138–2922 (–OH), and 1713–1669 (lactam carbonyl) cm⁻¹; ¹H NMR (200 MHz, CDCl₃) **5a** : δ 2.00 (t, 1H, *J* = 2.68 Hz, –CH₂–C≡C–H), δ 2.94 (d, 1H, *J* = 2.68 Hz, –CH₂–C≡C–H), and δ 2.96 (d, 1H, *J* = 2.68 Hz, –CH₂–C≡C–H) ppm **5b - 5e** : δ 1.81–1.83 (ABX, 1H, *J*_{AX}, *BX* = 1.47–2.69 Hz, –CH₂–C≡C–H), 2.67–2.92 (ABX, 1H, *J*_{AB} = 16.84–16.85, *J*_{AX} = 1.47–2.69 Hz, –CH₂–C≡C–H), and 2.93–3.08 (ABX, 1H, *J*_{AB} = 16.84–16.85, *J*_{BX} = 1.47–2.69 Hz, –CH₂–C≡C–H) ppm ; MS (M⁺ ion)].

Scheme 2

Table 1. Preparation of *N*-Alkyl-3-hydroxy-3-propargylisoindolin-1-ones **5a-e**

Product 5	Yield (%) ^{a)}	mp (°C)
5a	69	158-159
5b	83	111-112
5c	70	132
5d	83	160-161
5e	85	156

a) All yields are those of the isolated compounds.



a : R = H, **b :** R = Et, **c :** R = *n*-Pr

First, we tentatively carried out the ring enlargement reaction without isolation of **5b, c**. Namely, after reaction of **4b, c** with 1.2 mol eq. of propargylmagnesium bromide in Et₂O at 0°C for 5 min, 1.2 mol eq. of BF₃·OEt₂ was immediately added at 0°C and then the mixture was stirred at room temperature for 5 min. However, we could not obtain the desired 2-benzazepine-1,5-diones **6b, c** at all. Instead, unexpected bromodienes **7b, c** were obtained in 65% and 79% yields each as sole products. The stereochemistry (*E* or *Z*) of **7b, c** could not be determined by the ¹H NMR NOE experiment because of difficulty in assignment of the three olefinic proton signals. These compounds **7b, c** might be furnished by bromination onto plausible allenyl acyliminium species **9b, c** generated *in situ* from the corresponding ene-yne **8b, c**, which were formed by BF₃-promoted elimination of BrMgOH from the propargylmagnesium bromide adducts **5'b, c**, respectively. The detailed mechanistic study is now in progress.

Subsequently, the reaction conditions to obtain a satisfactory amount of 2-benzazepine-1,5-dione **6c** were examined by employing 3-hydroxy-3-propargylisoindolin-1-one **5c** and a range of 2.4 - 0.05 mol eq. of BF₃·OEt₂. Consequently, **6c** was best obtained in 56% yield in the presence of 0.15 mol eq. (15 mol %) of BF₃·OEt₂ in CH₂Cl₂ at -78°C over room temperature. Treatment of some other compounds **5a, b, d, e** with 15 mol% of BF₃·OEt₂ also resulted in 27 - 56% yields of 7-membered lactams **6a, b, d, e**, as shown in Table 2. Then, similar reactions of **5** toward **6** were examined in detail by exploiting slightly soft Lewis acids (15 mol%) such as "H₂B(CF₃SO₃)", "HB(CF₃SO₃)₂", "B(CF₃SO₃)₃", and "Al(CF₃SO₃)₃".⁵ Table 2 summarizes the experimental results.⁶

Table 2. Ring Enlargement Reaction of 3-Hydroxy-3-propargylisoindolin-1-ones **5** to 3,4-Olefinic 2-Benzazepine-1,5-diones **6**

Compd 5	Reaction Conditions			Product 6	Yield (%) ^{b)}	mp (°C)
	LA ^{a)}	Temp (°C)	Time (h)			
5a ^{c)}	B	-78 - r.t.	6.5	6a	27 ^{d)}	120 - 122
"	AT	-78 - 0	2.5	"	40 ^{e)}	"
"	BT	-78 - 10	2.6	"	81 ^{f)}	"
"	HBT	"	1.8	"	30 ^{g)}	"
5b	B	-78 - r.t.	6.5	6b	56	92 - 94
"	AT	-78 - 0	3	"	68	"
"	BT	-78 - 10	2.5	"	79	"
"	HBT	"	3	"	80	"
5c	B	-78 - r.t.	6.5	6c	56	73 - 74
"	AT	-78 - 0	6.5	"	51	"
"	BT	-78 - 10	3	"	78	"
"	HBT	"	3	"	98	"
"	H ₂ BT	"	1.5	"	93	"
5d	B	-78 - r.t.	6.5	6d	50	83 - 84
"	AT	-78 - 0	6.2	"	80	"
"	BT	-78 - 10	2.8	"	93	"
"	HBT	"	2.8	"	94	"
5e	B	-78 - r.t.	6.5	6e	54	125 - 126
"	AT	-78 - 0	3.2	"	71	"
"	BT	-78 - 10	4	"	79	"
"	HBT	"	3.5	"	85	"

a) LA : Lewis acid (15 mol%), B = BF₃•OEt₂, AT = "Al(CF₃SO₃)₃", BT = "B(CF₃SO₃)₃", HBT = "HB(CF₃SO₃)₂", H₂BT = "H₂B(CF₃SO₃)". b) All yields are those of isolated compounds. c) A solution (CH₂Cl₂ : THF = 2 : 1) was employed. d-g) Ene-yne compound **8a** was also obtained in various yields [d) 25%, e) 9%, f) trace, and g) 42%].

The desired ring enlargement reactions (**5** → **6**) with these slightly soft Lewis acids [approximate softness order⁷: "H₂B(CF₃SO₃)" > "HB(CF₃SO₃)₂" > "B(CF₃SO₃)₃" ≥ "Al(CF₃SO₃)₃" > BF₃] proved to be excellent in the yields of the 3,4-olefinic 2-benzazepine-1,5-diones **6b** - **e** except for compound **5a** with easy imine formation. These slightly soft Lewis acids may coordinate better with a soft Lewis base, "carbonyl oxygen atom" than with a hard base, "hydroxyl oxygen atom".⁷ Thus, these tentative Lewis acids were efficient for essential ring-opening of the γ -lactam moiety of **5** followed by intramolecular endo-mode cyclization between the resulting imidate moiety and the conjugated allenyl ketone moiety generated *in situ* as shown in Scheme 1. Treatment of **5c** with a hard acid reagent system, a mixture of BF₃•OEt₂ with CF₃SO₃H⁸ resulted in 57% yield of **6c**. Interestingly, with compound **5a**, ene-yne compound **8a** [mp 138-140 °C dec. (CH₂Cl₂-Et₂O)] was always produced as a by-product of **6a**. The structures of the synthesized 3,4-olefinic 2-benzazepine-1,5-diones **6a-e** were deduced from their characteristic spectroscopic data [IR (KBr) ν 1724-1713

(PhC \underline{O} CH=CR $_2$) and 1686-1575 (PhC \underline{O} NR $_2$) cm $^{-1}$; 1 H NMR (200 MHz, CDCl $_3$) δ 2.43-2.26 (s, 3H, =CR-CH $_3$) and 6.22-6.01 (s, 1H, -C \underline{O} CH=CR $_2$) ppm; MS (M $^+$ ion)].

In summary, we have established a new expeditious synthetic method for the 3,4-olefinic 2-benzazepine-1,5-diones based on the ring enlargement reaction of *N*-alkyl-3-hydroxy-3-propargylisoindolin-1-ones in the presence of a catalytic amount of the Lewis acid prepared from BH $_3$ •THF (or Me $_3$ Al) and CF $_3$ SO $_3$ H.

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References and Notes

1. a) Maruyama, K.; Kubo, Y. *Chem. Lett.*, **1978**, 851. b) Alonso, R.; Castedo, L.; Domínguez, D. *Tetrahedron Lett.*, **1986**, 27, 3539. c) Fang, F. G.; Danishefsky, S. J. *ibid.*, **1989**, 30, 2747. d) Busacca, C. A.; Johnson R. E. *ibid.*, **1992**, 33, 165 and references cited therein. e) Griesbeck, A. G.; Mauder, H. *Angew. Chem. Int. Ed. Engl.* **1992**, 31, 73. f) Lamas, C.; Saá, C.; Castedo, L.; Domínguez, D. *Tetrahedron Lett.*, **1992**, 33, 5653. g) Paleo, M. R.; Domínguez, D.; Castedo, L. *ibid.*, **1993**, 34, 2369 and references cited therein.
2. a) Nagao, Y.; Lee, W. S.; Kim, K. *Chem. Lett.*, **1994**, 389. b) Nagao, Y.; Lee, W. S.; Komaki, Y.; Sano, S.; Shiro, M. *ibid.*, **1994**, 597. c) Nagao, Y.; Lee, W. S.; Jeong, I.-Y.; Shiro, M. *Tetrahedron Lett.*, **1995**, 36, 2799.
3. Recent progress of the ring enlargement reactions, see : Hesse, M. "Ring Enlargement in Organic Chemistry", VCH publishers, Inc., New York, **1991**.
4. Gibson, M. S.; Bradshaw, R. W. *Angew. Chem. Int. Ed. Engl.* **1968**, 7, 919.
5. A solution prepared by treatment of BH $_3$ •THF or Me $_3$ Al / hexane with 1.3 - 3.3 mol eq. of CF $_3$ SO $_3$ H in CH $_2$ Cl $_2$ at 0 °C, was tentatively termed "H $_2$ B(CF $_3$ SO $_3$)", HB(CF $_3$ SO $_3$) $_2$ ", "B(CF $_3$ SO $_3$) $_3$ ", or "Al(CF $_3$ SO $_3$) $_3$ " by us.
6. A typical experimental procedure for the preparation of **6** : To a solution of BH $_3$ •THF (1M solution in THF) (32.7 μ l, 0.03 mmol) in anhydrous CH $_2$ Cl $_2$ (2 ml) was added CF $_3$ SO $_3$ H (6.3 μ l, 0.07 mmol) at 0 °C under N $_2$. The mixture was stirred at 0 °C for 10 min and then compound **5c** (50 mg, 0.2 mmol) was added at -78 °C. After being stirred at -78 over 10 °C for 3 h, the reaction mixture was submitted to the usual work-up and chromatographic purification (silica gel, ether : hexane = 1 : 2) to give compound **6c** (49 mg, 98% yield) as colorless needles from CH $_2$ Cl $_2$ - ether.
7. Ho, T. - L. "Hard and Soft Acids and Bases Principle in Organic Chemistry", Academic Press, New York, **1977**.
8. Cf. a) Nicolaou, K. C.; Hwang, C. - K.; Duggan, M. *J. Am. Chem. Soc.*, **1989**, 111, 6682. b) Jun, J. - G.; Ha, T. H.; Kim, D. - W. *Tetrahedron Lett.*, **1994**, 35, 1235.

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